This article was downloaded by: [University of Haifa Library]

On: 17 August 2012, At: 10:21 Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH,

UK



Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/gmcl19

Purely Organic Magnetism of Pyridyl-Substituted Stable Thioaminyl Radicals

Yozo Miura ^a , Shinya Kurokawa ^a , Masaaki Nakatsuji ^a & Yoshio Teki ^b

Version of record first published: 24 Sep 2006

To cite this article: Yozo Miura, Shinya Kurokawa, Masaaki Nakatsuji & Yoshio Teki (1999): Purely Organic Magnetism of Pyridyl-Substituted Stable Thioaminyl Radicals, Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals, 334:1, 195-204

To link to this article: http://dx.doi.org/10.1080/10587259908023317

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

^a Department of Applied Chemistry, Faculty of Engineering, Osaka City University, Sumiyoshi-ku, Osaka, 558-8585, Japan

^b Department of Material Science, Graduate School of Science, Osaka City University, Sumiyoshi-ku, Osaka, 558-8585, Japan

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Purely Organic Magnetism of Pyridyl-Substituted Stable Thioaminyl Radicals

YOZO MIURA^a, SHINYA KUROKAWA^a, MASAAKI NAKATSUJI^a and YOSHIO TEKI^b

^aDepartment of Applied Chemistry, Faculty of Engineering, Osaka City University, Sumiyoshi-ku, Osaka 558–8585, Japan and ^bDepartment of Material Science, Graduate School of Science, Osaka City University, Sumiyoshi-ku, Osaka 558–8585, Japan

N-[(4-Nitrophenyl)thio]- (1a) and N-[(2,4-dichlorophenyl)thio)]-2,6-diphenyl-4-(3-pyri-N-[(4-nitrophenyl)thio]dyl)phenylaminyl (1b),(2a) and N-[(2,4-dichlorophenyl)thio]-4-phenyl-2,6-di(3-pyridyl)phenylaminyl (2b) were generated by PbO₂ oxidation of the corresponding precursors and isolated as radical crystals. For 1b X-ray crystallographic analysis was performed. The magnetic susceptibility measurements were carried out for the isolated radicals using a SQUID magnetometer in the temperature range 1.8 - 300 K. The susceptibilities of 1a and 2a were analyzed in terms of a one-dimensional (1D) antiferromagnetic (AFM) regular Heisenberg model with exchange interaction of $2J/k_B = -63.4$ and -17.8K, respectively, and that of 1b was interpreted in terms of a 1D AFM alternating Heisenberg model with $2J/k_B = -12.8$ K and $\alpha = 0.91$. On the other hand, that of **2b** was explained in terms of a 1D ferromagnetic regular Heisenberg model with $2J/k_{\rm B} = 22.4$ K.

Keywords: Stable free radicals; Magnetism; Isolation; X-ray crystallography

INTRODUCTION

Since the discovery of the first purely organic ferromagnet with $T_c = 0.60 \text{ K}$ in 1991^[1], magnetism of stable free radical crystals has attracted much

attention. Since the discovery of the first organic ferromagnet, a variety of stable free radical crystals have been investigated, and 18 kinds of organic ferromagnets have been found^[2-13]. However, since free radicals are inherently unstable, examples of isolable free radicals are seriously limited[14], and the free radicals whose magnetism has been investigated have been almost limited to nitroxide and nitronyl nitroxides. Therefore, the quest of a new class of stable free radicals is strongly desired for the further advances in chemistry and physics of purely organic magnetism. Standing on this background, we have made an effort to search for a new class of isolable stable free radicals [15], and it has been found that N-(arylthio)-2,4,6triarylphenylaminyls and their derivatives are isolable stable free radicals^[16-19]. Magnetic studies of the thioaminyl radical crystals have shown that four radicals crystals couple ferromagnetically with $2I/k_B = +3.6 - +28.0 \text{ K}^{(20)}$.

In the present work five kinds of pyridyl-substituted thioaminyl radicals, 1a,1b, 2a, 2b, and 3 (see Chart 1) were investigated. Of the five radicals investigated, the former four radicals could be isolated as radical crystals, and one of them (1b) was investigated by X-ray crystallography. The magnetic studies for the isolated radical crystals revealed that, although the magnetic interactions of 1a, 1b, and 2a were antiferromagnetic (AFM), that of 2b was ferromagnetic (FM). Herein we report isolation, X-ray crystallographic analysis, and magnetic characterization of 1 and 2.

EXPERIMENTAL

Precursors 4, 5, and 6 were prepared according to Schemes 1, 2, and 3, and isolation of 1 and 2 were performed as follows: a benzene solution of a

precursor was treated with PbO₂. After filtration, the benzene was removed by freeze-drying, and the residue was crystallized from EtOH or hexane-ethyl acetate to give pure radical crystals.

Scheme 1

Scheme 2

RESULTS and DISCUSSION

Preparation of Precursors and Isolation of Radicals

Preparation of precursors 4, 5, and 6 are shown in Schemes 1, 2, and 3.

Oxidation of 4, 5, and 6 was carried out in benzene with PbO₂. When PbO₂ was added to a stirred solution of 4, 5, or 6, an initially light yellow $(4\mathbf{a}, 5\mathbf{a})$ or colorless solution $(4\mathbf{b}, 5\mathbf{b}, 6)$ immediately turned green $(1\mathbf{b}, 2\mathbf{a}, 2\mathbf{b})$, yellowish green $(1\mathbf{a})$, or purple (3), and intense ESR signals were observed from the solutions. Although 3 was gradually decomposed in solution $(t_{1/2} \sim 8 \text{ h at } 20 \,^{\circ}\text{C})$, $(1\mathbf{a}, 1\mathbf{b}, 2\mathbf{a}, 2\mathbf{b})$ were quite persistent and could be isolated radical crystals.

Isolation of 1a, 1b, 2a, and 2b were carried out as follows: the precursors were treated with PbO₂ in benzene and, after filtration, the solvent was removed by freeze-drying. The resulting dark green or blue crystalline residue was crystallized from hexane—ethyl acetate (1a, 1b, and 2b) or EtOH (2a) to give radical crystals in 19–38% yields. The IR spectra showed no presence of a NH group, and the elemental analyses agreed with the calculations. The magnetic susceptibility measurements carried out with a SQUID magnetometer showed >93% purity for the isolated radicals.

X-Ray Crystallographic Analysis

Although 1a, 1b, and 2b crystallized to microcrystals, 1b provided a sufficiently large single crystal for X-ray crystallography. The molecular and crystal structure of 1b are shown in Figures 1 and 2.

The X-ray crystallographic results show that there are two conformers in the crystals of **1b**. One is a syn-form, and the other is an anti-form. In the syn-form the pyridyl nitrogen and ortho chlorine atom on the phenylthiyl benzene ring are in the same side with respect to the ring A-N-S-ring E π -framework of the radical molecule, and in the anti-form they are in the opposite side to each other. The bond lengths and bond angles are nearly identical to each other, but the torsion angles are somewhat different.

Magnetic Characterization for Isolated Radicals

The magnetic properties of 1a, 1b, 2a, and 2b were investigated using polycrystalline samples in the temperature range 1.8–300 K on a SQUID magnetometer. The diamagnetic components were estimated using Pascal's constants. The radical purities of 1 and 2 were determined to be 98.5 (1a), 95.0 (1b), 93.0 (2a), and 99.0% (2b) from the magnetic susceptibility

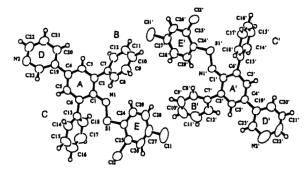


FIGURE 1. Molecular structure of 1b.

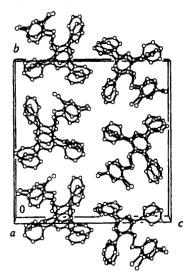


FIGURE 2. Crystal structure of 1b.

measurements.

Figure 3 shows the temperature dependence of $\chi_{mol}T$ for **2b**. The $\chi_{mol}T$ value are nearly constant in the temperature region above 50 K, and below this temperature it increases drastically with decreasing temperature. This profile indicates that the intermolecular magnetic interaction is FM, and the $\chi_{mol}T$ vs T plots have been well analyzed using a one-dimensional (1D) regular Heisenberg model (eq 1)^[21] with interchain interaction of $2I/k_B = 22.4$ K, as shown in Figure 3.

$$\mathcal{A} = -2J \sum S_i \cdot S_i \tag{1}$$

The temperature dependence of χ_{mol} for **1b** is shown in Figure 4. As found in the figure, χ_{mol} increases gradually and reaches a maximum at 10 K with decreasing temperature. Then, it decreases with decreasing temperature. This profile indicates that the interaction between the neighboring spins is AFM. The χ_{mol} vs T plots were analyzed with an alternating linear-chain model (eq 2, where α is an alternation parameter)^[22], and the best fit with the experiment gave $2J/k_B = -12.8$ K and $\alpha = 0.91$.

$$\mathcal{A} = -2J \sum (S_{2i-1} \cdot S_{2i} + aS_{2i} \cdot S_{2i+1})$$
 (2)

As found in Figure 2, the radical molecules are stacked alternately with the syn and anti-froms along the crystallographic *a* axis. The SOMO-SOMO overlap between the neighboring molecules is observed along this direction, and this leads to antiferromagnetic exchange interaction between the unpaired electron spins.

The temperature dependence of χ_{mol} for 1a are shown in Figures 5. χ_{mol} increases gradually and reaches a broad maximum at 40 K with decreasing temperature. Then, χ_{mol} decreases gradually with decreasing temperature, and below 20 K it again increases. This increase in χ_{mol} is probably due to the presence of isolated monoradicals which are randomly located in the lattice defects/or broken-chain edges. The contribution (x) of the

monoradical impurity (curve b) is estimated to be ca. 3.0% and the estimated Weiss constant (θ) is -2.0 K. The superposition of the theoretical curve (curve a) drawn by a 1D AFM regular Heisenberg model and curve b based on eq 3, where C is the Curie constant, gives curve c.

$$\mathcal{A} = -(1-x)2J \sum S_i \cdot S_j + x C/(T-\theta)$$
 (3)

The best-fit with the experiment gave $2J/k_B = -63.4$ K ($\alpha = 1$). A similar magnetic behavior was observed for 2a, and the best fit with the experiment using a 1D AFM regular Heisenberg model gave $2J/k_B = -17.8$ K. In this case paramagnetic impurity was nearly zero. Therefore, an increase in χ_{mol} in the low temperature region was not observed.

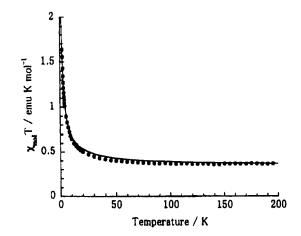
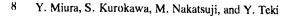


FIGURE 3. $\chi_{mol}T$ vsT plots for **2b**. The solid curve represents theoretical susceptibilities calculated with a 1D regular Heisenberg model with $2J/k_B = +22.4$ K.



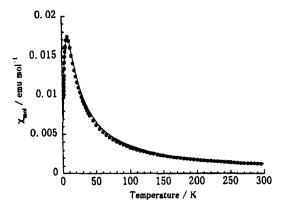


FIGURE 4. χ_{mol} vs T plots for 1b. The solid curve represents theoretical susceptibilities calculated with a 1D alternating Heisenberg model with $2J/k_B = -12.8$ K and $\alpha = 0.91$.

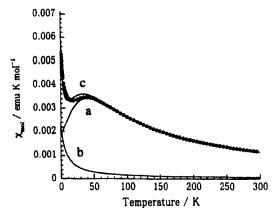


FIGURE 5. χ_{mol} vsT plots for 1a. Solid curve a represents theoretical susceptibilities calculated with 1D AFM regular Heisenberg model with $2J/k_B = -63.4$ K, and curve b is calculated for 3.0% of a monoradical impurity following the Curie-Weiss law with $\theta = -2.0$ K. Curve c is a superposition of curves a and b.

CONCLUSION

Five kinds of pyridyl-substituted thioaminyl radicals, **1a**, **1b**, **2a**, **2b**, and **3** have been generated, and the former four radicals could be isolated as radical crystals. The X-ray crystallographic analysis of **1b** has been performed and a large SOMO-SOMO overlap between the neighboring radicals was observed. The magnetic susceptibility measurements for the isolated radicals have showed that, although the intermolecular magnetic interactions of **1a** ($2J/k_B = -63.4$ K, a = 1.0), **1b** (12.8 K, 0.91), and **2a** (-17.8 K, 1.0) are antiferromagnetic, that of **2b** is ferromagnetic, and analysis of the susceptibility data with a 1D ferromagnetic regular Heisenberg model gave $2J/k_B = +22.4$ K.

Acknowledgments

The authors are grateful to Professor K. Matsumoto, Osaka City University, for valuable suggestions during X-ray crystallographic measurements.

References

- M. Kinoshita, P. Turek, M. Tamura, K. Nozawa, D. Shiomi, Y. Nakazawa, M. Ishikawa, M. Takahashi, K. Awaga, T. Inabe, and Y. Maruyama, *Chem. Lett.*, 1991, 1225.
- [2] R. Chiarelli, M.A. Novak, A. Rassat, and J.L. Tholence, Nature, 1993, 363, 147.
- [3] T. Nogami, K. Tomioka, T. Ishida, H. Yoshikawa, M. Yasui, F. Iwasaki, H. Iwamura, N. Takeda, and M. Ishikawa, Chem. Lett., 1994, 29.
- [4] T. Ishida, H. Tsuboi, T. Nogami, H. Yoshikawa, M. Yasui, F. Iwasaki, H. Iwamura, N. Takeda, and M. Ishikawa, Chem. Lett., 1994, 919.
- [5] T. Nogami, T. Ishida, H. Tsuboi, H. Yoshikawa, H. Yamamoto, M. Yasui, F. Iwasaki, H. Iwamura, N. Takeda, and M. Ishikawa, Chem. Lett., 1995, 635.
- [6] T. Nogami, T. Ishida, M. Yasui, F. Iwasaki, N. Tekeda, M. Ishikawa, T. Kawakami, and K. Yamaguchi, Bull. Chem. Soc. Jpn., 69, 1841 (1996).
- [7] K. Togashi, R. Imachi, K. Tomioka, H. Tsuboi, T. Ishida, T. Nogami, N. Takeda, and M. Ishikawa, Bull. Chem. Soc. Jpn., 69,2821 (1996).
- [8] M.M. Matsushita, A. Izuoka, T. Sugawara, T. Kobayashi, N. Wada, N. Takeda, and M. Ishikawa, J. Am. Chem. Soc., 119, 4369 (1997)..
- [9] J. Cirujeda, M. Mas, E. Molins, F.L. de Panthou, J. Laugier, J.G. Park, C. Paulsen, P. Rey, C. Rovira, and J. Veciana, J. Chem. Soc., Chem. Commun., 1995, 709.
- [10] A. Caneschi, F. Ferrano, D. Gatteschi, A. Lirzin, M.A. Novak, E. Rentschler, and R. Sessoli, Adv. Mater., 7, 476 (1995).
- [11] K. Mukai, K. Konishi, K. Nedachi, and K. Takeda, J. Phys. Chem., 100, 9658 (1996).
- [12] K. Takeda, T. Hamano, T. Kawae, M. Hidaka, M. Takahashi, S. Kawasaki, and K. Mukai, J. Phys. Soc. Jpn., 64, 2343 (1995).
- [13] T. Sugimoto, M. Tsuji, T. Suga, N. Hosoito, M. Ishikawa, N. Takeda, M. Shiro, Mol. Cryst. Liq. Cryst., 272, 183 (1995).

- [14] (a) A.R. Forrester, J.M. Hay, and R.H. Thomson, Organic Chemistry of Stable Free Radicals (Academic Press, London and New York, 1968); (b) E.G. Rozantsev, Free Nitroxide Radicals (Prenum Press, New York and London, 1970) (c) L.B. Volodarsky, V.A. Reznikov, and V.I. Ovcharenko, Synthetic Chemistry of Stable Nitroxides (CRC Press: Boca Raton, 1994).
- [15] Y. Miura, Tren. Org. Chem., 6, 197 (1997); Y. Miura, Recent Res. Devel. in Org. Chem., 2, 251 (1998).
- [16] (a) Y. Miura and A. Tanaka, J. Chem. Soc., Chem. Commun., 1990, 441; (b) Y. Miura, A. Tanaka, and K. Hirotsu, J. Org. Chem., 56, 6638 (1991); (c) Y. Miura, Y. Kitagishi, and S. Ueno, Bull. Chem. Soc. Jpn., 67, 3282 (1994).
- [17] (a) Y. Miura, E. Yamano, A. Tanaka, and Y. Ogo, Chem. Lett., 1992, 1831; (b) Y. Miura, E. Yamano, and A. Tanaka, J. Org. Chem., 59, 3294 (1994); (c) Y. Miura and E. Yamano, J. Org. Chem., 60, 1070 (1995); (d) Y. Miura, H. Oka, and E. Yamano, Y. Teki, T. Takui, and K. Itoh, Bull. Chem. Soc. Jpn., 68, 1187 (1995).
- [18] Y. Miura, T. Fuchikami, and M. Momoki, Chem. Lett., 1994, 2127; (b) Y. Miura, M. Momoki, T. Fuchikami, Y. Teki, and H. Mizutani, J. Org. Chem., 61, 4300 (1996).
- [19] Y. Miura, M. Momoki, M. Nakatsuji, and Y. Teki, J. Org. Chem., 63, 1555 (1998).
- [20] Y. Teki, K. Itoh, A. Okada, H. Yamakage, T. Kobayshsi, K. Amaya, S. Kurokawa, S. Ueno, and Y. Miura, Chem. Phys. Lett., 270, 573 (1997).
- [21] J.C. Bonner and M.E. Fisher, Phys. Rev., A135, 640 (1964).
- [22] (a) W. Duffy, Jr. and K.P. Barry, Phys. Rev., 165, 647 (1968); (b) J.W. Hall, W. March, R.R. Weller, and W.E. Hatfield, Inorg. Chem., 20, 1033 (1981).